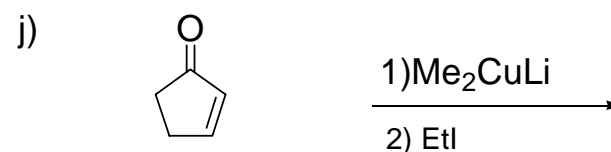
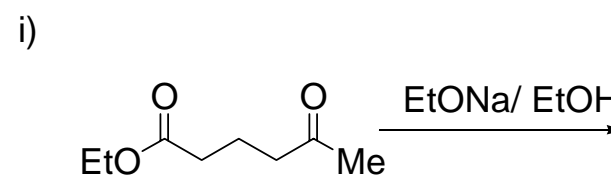
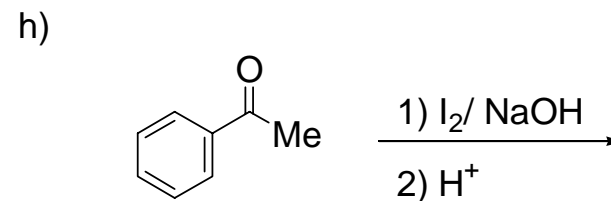
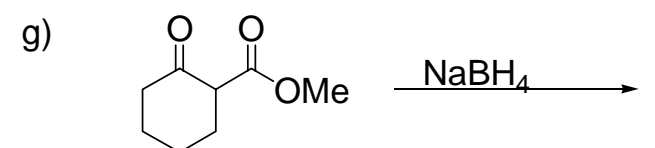
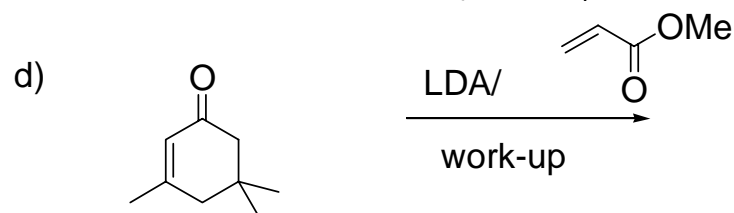
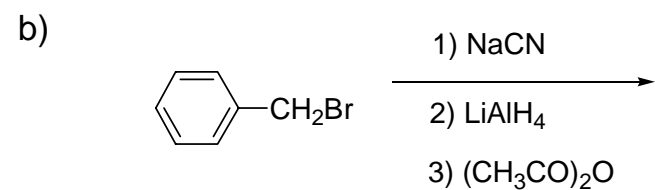
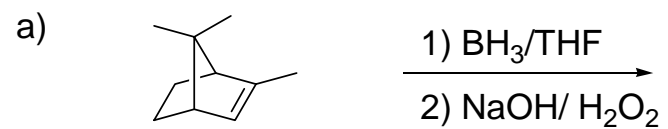


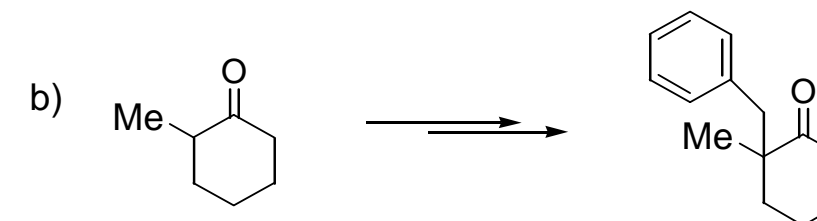
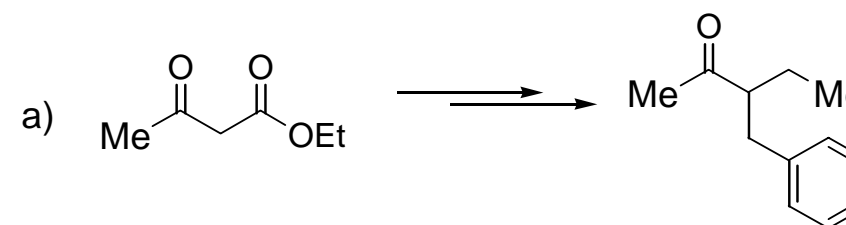
國立嘉義大學九十四學年度
應用化學系碩士班招生考試試題

科目：綜合化學 I

1. Provide the likely product(s) with proper stereochemistry and indicate the major one if more than one product formed. (3 % for each question)



2. Please complete the following synthesis, starting from the given material. (10 % for each question)



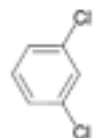
背面尚有試題

3. Arrange the order by putting “>” or “<” between the following molecules. (10%)

- | | | |
|-----------------------------------|-------------------------------|--|
| (a) OSF ₂ | OSCl ₂ | OSBr ₂ (halogen-S-halogen angle) |
| (b) NH ₃ | PH ₃ | AsH ₃ (H-N-H angle) |
| (c) H ₂ O | H ₂ S | H ₂ Se (H-O-H angle) |
| (d) PF ₃ | PCl ₃ | PBr ₃ (X-P-X angle) |
| (e) ClO ₃ ⁻ | BrO ₃ ⁻ | IO ₃ ⁻ (O-X-O angle) |

4. Give the point group of the following: (10 %)

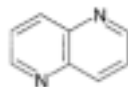
(a)



(b)



(c)



(d)



5. Predict the order of acidity between BF₃, BCl₃, BBr₃ and **explain why**. (10%)

6. What's the coordination number of the following structures. (10 %)

- | | |
|-----------------------------------|-------|
| (a) a primitive cubic structure | _____ |
| (b) a body-centered-cubic lattice | _____ |
| (c) a face-centered-cubic lattice | _____ |
| (d) a cesium chloride lattice | _____ |
| (e) a rock-salt lattice | _____ |

7. Give the synthesis for *cis* and *trans*-[Pt(NH₃)₂Cl₂], starting with [Pt(NH₃)₄]⁺² or [PtCl₄]⁻². (10 %)