## 國立嘉義大學 99 學年度

# 應用化學系碩士班(甲組)招生考試試題

### 科目:綜合化學 [

一、有機化學(50分)

1. Please complete the following reactions by giving the major product, with correct stereochemistries if applicable. (30 points, 3 points for each reaction)



3. Pleaes answer the following questions; (10 points) a) An unknown compound with formula  $C_4H_8Cl_2$ , Please solve the structure using the following <sup>1</sup>HNMR spectrum. (5 points)



b) Please explain the following observations;



- (1) Draw the corresponding products. (3 points)
- (2) Why there is difference in reaction rates between these two reactions ? (2 points)

2. Please complete the following syntheses, starting from the giving starting material. Write out all the needed reagents and transformed products. (10 points, 5 points for each synthesis)

a)





two products

#### only one product

- 二、無機化學(50分)
- 1. Using the Molecular Orbital to calculate the bond order of the following molecules:  $O_2^+$ ,  $O_2^{-1}$ ,  $B_2$ , NO and predict which molecule(s) is(are) paramagnetic? (8points)



- 2. Draw the three possible geometric structures of  $BrF_3$  and explain which is the most favorable structure? (6 points):
- 3. The  $4f_z(x^2-y^2)$  orbital has the angular function Y=(constant)z(x^2-y^2). (a) Hhow many angular nodes does it have? Describe the angular nodal surfaces.
  - (b) How many spherical nodes does it have? Explain your reasons. (8 points)

4. Choose and explain: (6 points)

(a) Strongest Bronsted acid :	Si	nH <sub>4</sub> St	oH <sub>3</sub> T	eH <sub>2</sub>	
(b) Strongest Base to $H^+$ (gas phase):	$NH_3$	CH <sub>3</sub> NI	H <sub>2</sub> (CH	(3) <sub>2</sub> NH	$(CH_3)_3N$
(c) Strongest Base to BMe <sub>3</sub> :	Pyridine	2-methy	lpyridine	4-met	hylpyridine

5. Predict the number of unpaired electrons for each of the following: (8 points) (a) a tetrahedral  $d^6$  ion (b)  $[Co(H_2O)_6]^{2+}$ (c)  $[Cr(H_2O)_6]^{3+}$ (d) a square-planar  $d^7$  ion

6. Predict the *MAJOR* product of the following reactions: (8 points)

 $Mo(CO)_6$  +  $Ph_2P CH_2 PPh_2$ 

$$(\eta^5 - C_5H_5)(\eta^1 - C_3H_5)Fe(CO)_2$$



- 7. Which of the complexes has the **lowest** energy of CO stretching frequency in the IR region(6 points)? give the explainations.
  - (a)  $[Mn(CO)_6]^+$  $Cr(CO)_6$ (b)  $Mo(CO)_3(PCl_3)_3$   $Mo(CO)_3(PCl_2Ph)_3$   $Mo(CO)_3(PMe_3)_3$





С

 $[V(CO)_6]^{-1}$