

國立嘉義大學九十五學年度
光電暨固態電子研究所碩士班招生考試試題

科目：近代物理

Coulomb constant $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N}\cdot\text{m}^2/\text{C}^2$; Planck's constant $h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}$

Rest mass of electron $m_e = 9.1 \times 10^{-31} \text{ kg}$; Rest mass of proton $m_p = 1.67 \times 10^{-27} \text{ kg}$

第一部分：計算題

- The atom of positronium consists of an electron and a positron orbiting about each other.
 - Find the Bohr radius of this system. (8%)
 - Find the longest wavelength of the photon released in the transition to $n = 1$. (7%)
- Consider a system of three noninteracting, identical spin 1/2 particles that are in the same spin state $\left| \frac{1}{2}, \frac{1}{2} \right\rangle$ and confine to move in a one-dimensional infinite potential well of length a : $V(x) = 0$ for $0 < x < a$ and $V(x) = \infty$ for other values of x . Determine the energy and wave function of the ground state. (15%)
- Write down the quantum number for the states described in spectroscopic notation as : ${}^2S_{3/2}$, 5P_3 , 3D_2 . (8%)
 - Determine these states are possible or impossible, and explain why. (7%)
- Considering a collision between a photon with energy and momentum (E_0, p_0) and a free stationary electron, derive the Compton equation by means of the quantum explanation and relativistic theory. (8%)
 - In what region of the electromagnetic spectrum will be dominated by the Compton scattering, please specify why? (Compton wavelength = 0.0243 \AA) (7%)
- The optical electron configuration of an atom with two optically active electrons is $3d^2$. The energy levels of the atom are in spin-orbit (LS) interactions.
 - Find the possible values of quantum numbers s' , ℓ' and j' , where s' , ℓ' and j' are total spin, total orbital and total atomic angular momentum quantum numbers, respectively. (8%)
 - Indicate the splitting of the energy levels by a typical LS coupling configuration. You have to label quantum numbers and spectroscopic notations of the energy levels. (7%)

第二部分：解釋名詞

- Wien displacement law (5%)
- De Broglie wavelength (5%)
- Paschen-Bach effect (5%)
- Zeeman effect (5%)
- Rayleigh scattering (5%)